



Hyperpolarization of *cis*- $^{15}\text{N}_2$ -Azobenzene by Parahydrogen at Ultralow Magnetic Fields**

Kirill F. Sheberstov,^{*,[a, b]} Vitaly P. Kozinenko,^[c, d] Alexey S. Kiryutin,^[c, d] Hans-Martin Vieth,^[e] Herbert Zimmermann,^[f] Konstantin L. Ivanov^{+, [c, d]} and Alexandra V. Yurkovskaya^[c, d]

Dedicated to the memory of Konstantin L'vovich Ivanov (1977–2021).

The development of nuclear spins hyperpolarization, and the search for molecules that can be efficiently hyperpolarized is an active area in nuclear magnetic resonance. In this work we present a detailed study of SABRE SHEATH (signal amplification by reversible exchange in shield enabled alignment transfer to heteronuclei) experiments on $^{15}\text{N}_2$ -azobenzene. In SABRE SHEATH experiments the nuclear spins of the target are hyperpolarized through transfer of spin polarization from parahydrogen at ultralow fields during a reversible chemical process. Azobenzene exists in two isomers, *trans* and *cis*. We show that all nuclear spins in *cis*-azobenzene can be efficiently

hyperpolarized by SABRE at suitable magnetic fields. Enhancement factors (relative to 9.4 T) reach up to 3000 for ^{15}N spins and up to 30 for the ^1H spins. We compare two approaches to observe either hyperpolarized magnetization of $^{15}\text{N}/^1\text{H}$ spins, or hyperpolarized singlet order of the ^{15}N spin pair. The results presented here will be useful for further experiments in which hyperpolarized *cis*- $^{15}\text{N}_2$ -azobenzene is switched by light to *trans*- $^{15}\text{N}_2$ -azobenzene for storing the produced hyperpolarization in the long-lived spin state of the ^{15}N pair of *trans*- $^{15}\text{N}_2$ -azobenzene.

1. Introduction

Parahydrogen can be used to hyperpolarize nuclear spins allowing enhancement of signals in nuclear magnetic resonance (NMR) and magnetic resonance imaging (MRI) by up to 5 orders

of magnitude.^[1,2] This method is applicable when a pairwise hydrogenation reaction occurs; the singlet spin order (population imbalance between the singlet and triplet states of the spin pair) of parahydrogen can be transferred into and redistributed across the nuclear spin system of the target molecule. Of particular interest are reversible hydrogenation reactions happening in catalytic complexes: in this case a dihydrogen molecule H_2 that is dissolved in a solution can reversibly bind to a metal centre. It is possible to transfer its nuclear spin order to another target molecule that binds to the same complex simultaneously with the hydrogens. The method is called SABRE (Signal Amplification By Reversible Exchange),^[3] and it enables the polarization of a wide variety of substrates,^[4] including important biomarkers that can be used as contrast agents for bio-imaging.^[5] Importantly, this technique allows for many repetitions of the hyperpolarization process with the same target molecules since all the chemical processes are cyclic.

The search for new targets for SABRE is challenging, as it is not easy to predict in advance the molecules that are suitable. The catalytic complexes used for SABRE are usually based on iridium (III), thus the target molecules should be able to reversibly bind to iridium. Therefore, the molecule must contain an electron-donating atom, such as nitrogen. There are currently more than a hundred compounds known that can be hyperpolarized by SABRE, and most of them contain a nitrogen atom.^[6]

The most efficient way to transfer the spin order of parahydrogen to the target molecule exploits a coherent transfer through the *J*-coupling network; this process depends strongly on the magnetic field at which SABRE is performed. Heteronuclear spins such as ^{15}N and ^{13}C can be hyperpolarized

[a] Dr. K. F. Sheberstov
Institut für Physik, Johannes Gutenberg Universität-Mainz
55128 Mainz, Germany
E-mail: sheberst@uni-mainz.de

[b] Dr. K. F. Sheberstov
Helmholtz-Institut Mainz, GSI Helmholtzzentrum für Schwerionenforschung
55128 Mainz, Germany

[c] V. P. Kozinenko, Dr. A. S. Kiryutin, Prof. K. L. Ivanov,⁺ Dr. A. V. Yurkovskaya
International Tomography Center SB RAS
Novosibirsk, 630090, Russia

[d] V. P. Kozinenko, Dr. A. S. Kiryutin, Prof. K. L. Ivanov,⁺ Dr. A. V. Yurkovskaya
Novosibirsk State University
Novosibirsk, 630090, Russia

[e] Prof. H.-M. Vieth
Freie Universität Berlin, Fachbereich Physik
14195 Berlin, Germany

[f] H. Zimmermann
Department of Biomolecular Mechanisms, Max-Planck-Institut für Medizinische Forschung
69120 Heidelberg, Germany

[⁺] Deceased

[**] A previous version of this manuscript has been deposited on a preprint server: <https://arxiv.org/abs/2103.00289>



Supporting information for this article is available on the WWW under <https://doi.org/10.1002/cphc.202100160>



An invited contribution to a Special Collection on Parahydrogen Enhanced Resonance



© 2021 The Authors. ChemPhysChem published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution Non-Commercial License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited and is not used for commercial purposes.

by SABRE in zero- to ultralow-fields (ZULF). In ZULF the difference between the Larmor frequency of protons and the heteronuclear spin is comparable to the J -coupling between them, making spontaneous coherent polarization transfer through J -couplings possible. This method is called SABRE SHEATH (shield enables alignment transfer to heteronuclei).^[7,8]

Relaxation times of ^{15}N and ^{13}C spins are typically much longer than those of ^1H spins, making it possible to store the hyperpolarization up to several minutes. An alternative possibility is to store hyperpolarization in a so-called long-lived state (LLS),^[9] a state that is immune to certain relaxation mechanisms. The longest relaxation times are observed for singlet order formed within pairs of ^{15}N spins^[2,10–14] or ^{13}C spins,^[15] and nonequilibrium spin states can exist for up to tens of minutes in liquids at room temperature. The possibility to store hyperpolarization expands the field of applications and has intensified the development of SABRE experiments producing LLS.^[13,14,16,17] Here we report on SABRE SHEATH experiments with $^{15}\text{N}_2$ -azobenzene (ABZ), a molecular target with LLS properties.

ABZ exists in two isomeric forms, *trans*-ABZ and *cis*-ABZ, and the prevailing form in solution can be controlled by light (see Figure 1a). Recently, it was discovered that the singlet state of the ^{15}N spin pair in *trans*- $^{15}\text{N}_2$ -ABZ is long-lived, which enables sustaining non-thermal spin order for up to 50 minutes even at high magnetic field ($> 10\text{ T}$).^[12] An illustration of the SABRE process of ABZ is shown in Figure 1b.

The possibility to hyperpolarize *cis*- $^{15}\text{N}_2$ -ABZ with SABRE was reported previously,^[4] however it was a large scale study of many compounds, information about SABRE of $^{15}\text{N}_2$ -ABZ was limited, and the polarization level was not determined accurately. Here we present a detailed study on SABRE SHEATH of $^{15}\text{N}_2$ -ABZ and compare experiments, in which hyperpolarized magnetization of ^{15}N and ^1H is observed to experiments in which hyperpolarized singlet order of the ^{15}N spins pair is observed.

Experimental Section

The precatalyst, $[\text{Ir}(\text{IMes})(\text{COD})\text{Cl}]$ (COD stands for 1,5-cyclooctadiene), was synthesized using the procedure described in reference^[18] and ^{15}N -enriched ABZ was synthesized as described in



Konstantin L'vovich Ivanov passed away on 5th of March, 2021 at the age 44, during the review process of this manuscript. In his early career, he made major contributions to the theory of chemical reaction kinetics in liquid phase. Later, he contributed to unravelling mechanisms of light-induced nuclear hyperpolarization in liquids and solids using the concept of level anti-crossing (LAC). In recent years, his main research efforts, both theoretical and experimental, were concentrated on spin and chemical dynamics in PHIP and SABRE methods using LAC; for this, he was awarded the Günther Laukien Prize in 2020. The work on SABRE of azobenzene was one of the last of his project.

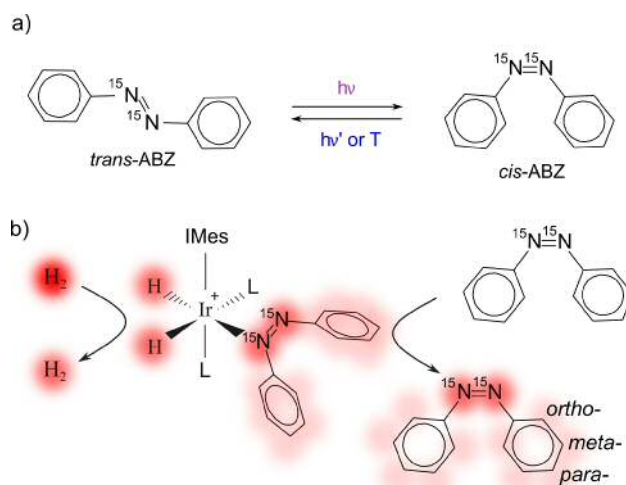


Figure 1. a) Chemical structure and photochromic property of azobenzene (ABZ). b) Scheme of SABRE with *cis*-ABZ showing polarization transfer from a parahydrogen molecule to *cis*-ABZ within a reversibly formed Ir-IMes activated complex. IMes stands for 1,3-dimesitylimidazol-2-ylidene, L stands for another possible ligands in the complex, either azobenzene, methanol, or Cl^- .

reference.^[19] Deuterated methanol- d_4 (99.8% atom %D) was purchased from Carl Roth.

The samples consisted of 50 mM of $^{15}\text{N}_2$ -ABZ and of 3 mM of the precatalyst, $[\text{Ir}(\text{IMes})(\text{COD})\text{Cl}]$ dissolved in deuterated methanol. The sample was irradiated by near UV light ($\sim 380\text{ nm}$) to convert *trans*- $^{15}\text{N}_2$ -ABZ into *cis*- $^{15}\text{N}_2$ -ABZ. After irradiation, the *cis*- $^{15}\text{N}_2$ -ABZ to *trans*- $^{15}\text{N}_2$ -ABZ ratio reached 1:1. Next, the parahydrogen gas was bubbled through the solution to activate the $[\text{Ir}(\text{IMes})(\text{COD})\text{Cl}]$ complex. This process was performed inside the NMR spectrometer and monitored by ^1H NMR. The activation of the complex took up to 10 minutes without addition of any substrates except for $^{15}\text{N}_2$ -ABZ. The temperature in experiments was set to 20°C .

Chemical shifts of ^{15}N are referenced on the Ξ -scale utilizing the absolute Larmor frequency of TMS and $\Xi_{\text{N}} = 10.1329118$ for liquid ammonia to obtain the absolute frequency for 0 ppm of the ^{15}N spins.^[20]

Parahydrogen was enriched up to 90% using a Bruker parahydrogen generator, and a pressure of 3 bar was used for all experiments. The bubbling time τ was optimized by measuring the “build up curve”, where the intensity of the hyperpolarized signals is plotted as function of τ (not shown here). After bubbling in a magnetic field of 500 nT for 30 s, the intensity reached a plateau with the level of hyperpolarization about 1.7 times higher as compared to a bubbling time of 10 s. For all subsequent experiments the bubbling period was fixed at $\tau = 10\text{ s}$ – this made it possible to achieve a high degree of polarization with a high experiment repetition rate. Further details of the experimental setup are given in reference.^[21]

The spin system of *cis*- $^{15}\text{N}_2$ -ABZ includes 12 coupled spins; all of them can be hyperpolarized by SABRE. There are slightly different experimental protocols allowing different ways to observe the hyperpolarization. These methods are presented in Figure 2. Most of the steps are identical: the sample containing $^{15}\text{N}_2$ -ABZ and precatalyst was placed in a magnetic shield; the required ultralow field was set up using a set of coils, and the parahydrogen gas was bubbled through the solution. The supply of gas was then stopped

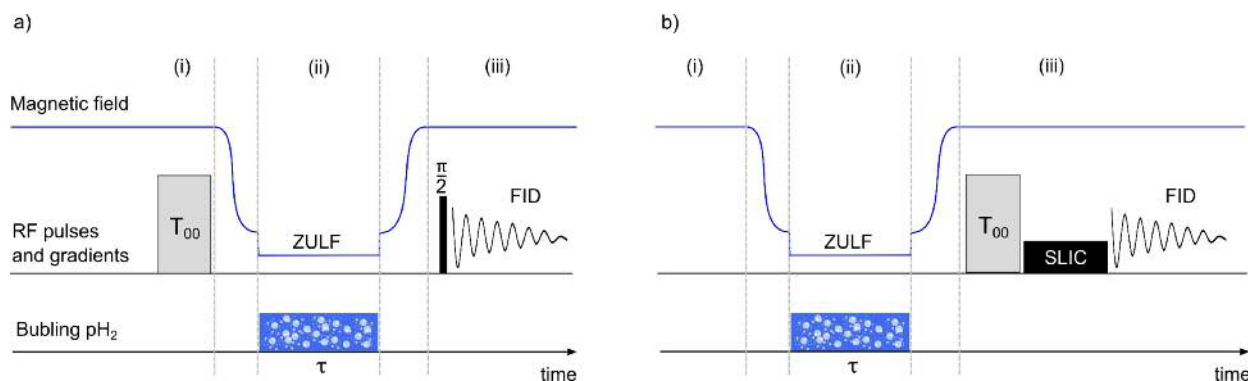


Figure 2. Experimental protocols: a) Field cycling in a SABRE SHEATH experiment with consecutive (i) destruction of equilibrium magnetization at the high field of the NMR spectrometer, followed by transfer of the sample into the magnetic shield, (ii) bubbling parahydrogen at ZULF conditions in a variable magnetic field for a time τ , and (iii) return to high field and excitation of the NMR signals by a hard pulse. b) Field cycling in a SABRE SHEATH experiment to observe hyperpolarized singlet order of the ^{15}N spin pair in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. In this case, excitation is performed by using a SLIC pulse (see text). The T_{00} block indicates a filter based on 90° pulses with pulsed field gradients; for details see reference [23].

by immediate equilibration of pressure at the inlet and outlet of the NMR tube at a flow rate about 40 sccm; then the magnetic field was instantaneously switched to 25 μT . Such an instantaneous jump of the magnetic field is not necessary, but it allowed better reproducibility of experiments. Then the sample was transferred inside the NMR spectrometer (9.4 T, 400 MHz ^1H Larmor frequency) where different detection sequences were applied.

The first possibility is to apply a hard 90° pulse (Figure 2a), detecting magnetization of the hyperpolarized spins. Because the spin system of $^{15}\text{N}_2\text{-ABZ}$ contains both ^{15}N and ^1H spins, detection can be performed either by using the ^{15}N or the ^1H channel.

The second possibility is to detect hyperpolarized singlet order of the ^{15}N spin pair in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. It is worth noting that the two ^{15}N spins in $^{15}\text{N}_2\text{-ABZ}$ are close to magnetic equivalence, the eigenstates of this pair are close to the singlet and triplet states, hence in most cases singlet order is not observable after applying the usual hard NMR pulses. Established techniques for converting singlet order into magnetization; here we use a method that is called “spin-lock induced crossing” (SLIC).^[22] SLIC pulses were adjusted to convert singlet order of the ^{15}N spin pair of $^{15}\text{N}_2\text{-ABZ}$ into observable magnetization.^[12,19] SLIC can be used to convert the singlet order into either ^{15}N magnetization or ^1H magnetization of the *ortho*-protons in $^{15}\text{N}_2\text{-ABZ}$. Consequently, the SLIC pulse was applied either at the ^{15}N resonance frequency (525 ppm for $\text{cis-}^{15}\text{N}_2\text{-ABZ}$) or at the $^1\text{H-ortho}$ frequency (6.85 ppm for $\text{cis-}^{15}\text{N}_2\text{-ABZ}$). The nutation frequency matched the $^1J_{\text{NN}}$ -coupling, which is 22 Hz, and the pulse duration was 0.3 s.

2. Results and Discussion

Polarization transfer in the spin network of $^{15}\text{N}_2\text{-ABZ}$ is a sophisticated process. Hyperpolarization propagates through the strongly coupled nuclei eventually polarizing remote spins. As we show further, the protons in the phenyl rings of $^{15}\text{N}_2\text{-ABZ}$ can be polarized only when the SABRE process is performed at ultralow fields. This is similar to previous studies of metronidazole where remote spins are hyperpolarized through the J -coupling network.^[24] Also, different spin orders are polarized; of particular interest are ^{15}N magnetization and singlet order of

the ^{15}N spin pair. The ability to hyperpolarize singlet order is reminiscent of the performance of SABRE SHEATH methods in diazarinis,^[16,17] diazines,^[14] and pyridazine derivatives.^[13]

The large size of the spin system makes it difficult to model the experiments: the full spin system in the complex with one $^{15}\text{N}_2\text{-ABZ}$ molecule contains 14 coupled spins. Nevertheless, some of the details can be understood from exploring the magnetic field dependencies of SABRE and by modeling the process with simplified spin systems.

The relevant J -couplings in the Ir-IMes complex with $^{15}\text{N}_2\text{-ABZ}$ are shown in Figure 3a, while the J -couplings in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ are shown in Figure 3b. The values for free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ are taken from reference,^[19] while the J -couplings in Ir-IMes complex were determined from the line splittings in ^1H and ^{15}N NMR spectra. These spectra were acquired after bubbling parahydrogen in high field at 5°C , which slowed down the exchange and made the splitting visible (see Supporting Information).

Cis and *trans* $^2J_{\text{HN}}$ -couplings were distinguished in accordance with previous studies of $^2J_{\text{HN}}$ -couplings in Rh-based complexes.^[25] Interestingly, the $^1J_{\text{NN}}$ -coupling in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ is different for the free form (~ 22 Hz) and for the molecule bound in equatorial position (~ 18 Hz). This is because the electron density is shifted from the attached site towards the metal center.

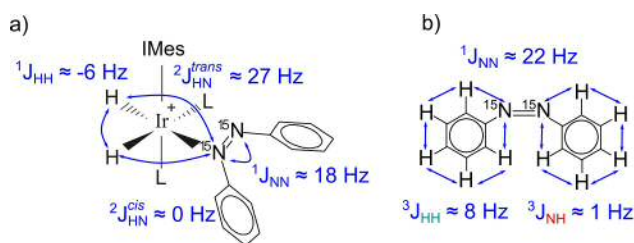


Figure 3. J -couplings in the Ir-IMes complex (a) and in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ (b). IMes stands for 1,3-dimesitylimidazol-2-ylidene, L stands for another possible ligands in the complex, either azobenzene, methanol, or Cl^- .

2.1. Magnetization Formed by SABRE SHEATH

A typical ^{15}N spectrum obtained after SABRE SHEATH of $^{15}\text{N}_2$ -ABZ at 500 nT (protocol of Figure 2a) is shown in Figure 4. In this figure the strongest signal near 525 ppm corresponding to the free *cis*- $^{15}\text{N}_2$ -ABZ is cut. There are several other hyperpolarized signals, which otherwise are invisible in the thermal equilibrium spectrum. All ^{15}N nuclei in the sample solution belong to $^{15}\text{N}_2$ -ABZ, so all these signals are attributed to different signals of either *cis*- or *trans*- $^{15}\text{N}_2$ -ABZ attached in equatorial or axial sites of the Ir-IMes complex. Detailed assignment of these signals was not our goal here, but we determined the signals corresponding to *cis*- $^{15}\text{N}_2$ -ABZ attached to Ir-IMes in equatorial position. These signals are highlighted in Figure 4. They are broadened due to chemical exchange. The signal at ~ 420 ppm corresponds to the ^{15}N position coordinated with the Ir, whereas the remote ^{15}N nitrogen has almost the same chemical shift as in the free *cis*- $^{15}\text{N}_2$ -ABZ. The assignment was confirmed by correlation experiments showing that there is a $^2J_{\text{HN}}$ -coupling between the attached ^{15}N and one of the parahydrogen protons in the Ir-IMes complex as well as

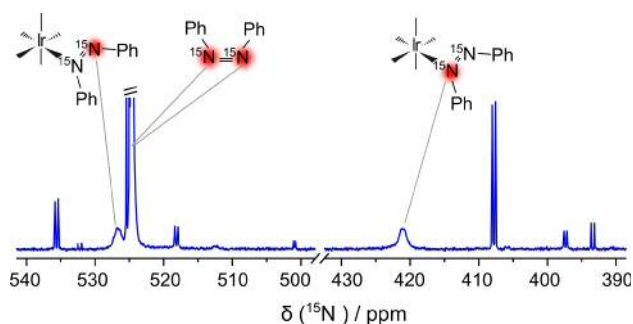


Figure 4. Fragments of the ^{15}N NMR spectrum showing signals of $^{15}\text{N}_2$ -ABZ in Ir-IMes complexes, polarized via SABRE SHEATH at 500 nT. The phase of the spectrum was adjusted to have absorptive signals. The experimental protocol is shown in Figure 2a.

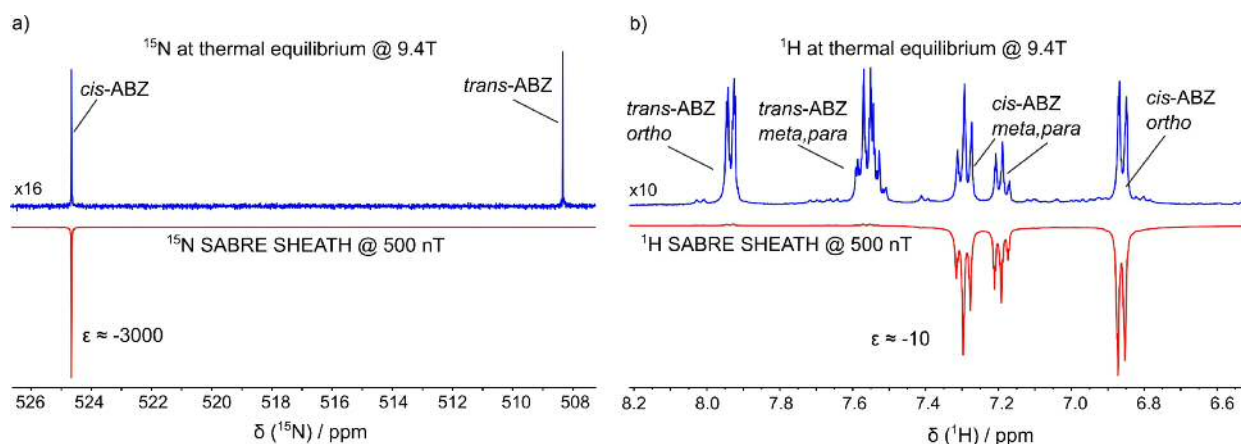


Figure 5. ^{15}N NMR spectra (a) and ^1H NMR spectra (b) showing comparison of $^{15}\text{N}_2$ -ABZ signals in thermal equilibrium (blue, absorptive peaks) and after SABRE SHEATH hyperpolarization at 500 nT (red, emissive peaks). The phases in SABRE SHEATH spectra were set with respect to corresponding thermal spectrum. The ^{15}N NMR spectrum at thermal equilibrium shown in (a) was summed for 128 times. Otherwise, all the compared spectra were acquired with identical acquisition and processing parameters. The experimental protocol for the SABRE experiments is shown in Figure 2a.

correlation between the ^{15}N signals (see Supporting Information).

Only the free form of *cis*- $^{15}\text{N}_2$ -ABZ and not the *trans*-ABZ is hyperpolarized as can be concluded from comparison of the thermal and the hyperpolarized spectra shown in Figure 5. Probably, absence of the hyperpolarization for the free *trans*- $^{15}\text{N}_2$ -ABZ can be explained by steric restrictions. For the shown spectrum (Figure 4 and Figure 5a) the enhancement factor was $|\epsilon| \approx 3000$. It was calculated as ratio between the integral intensity of the SABRE SHEATH spectrum and the intensity observed in the thermal equilibrium polarized spectrum. The same sample was used to acquire both spectra, but the thermally polarized spectrum is sum of 128 acquisitions to get an acceptable signal-to-noise ratio (SNR).

^1H signals of *cis*- $^{15}\text{N}_2$ -ABZ were also hyperpolarized (Figure 5b). The enhancement factors were $|\epsilon| \approx 10$ for each of the proton groups. This is two orders of magnitude less than the enhancement factors for the ^{15}N signal. The difference can be partially explained by the fact that ^1H spins have a ~ 10 times higher absolute value of the gyromagnetic ratio, and therefore thermal polarization of protons is ~ 10 times larger compare to nitrogen nuclei. Another order of magnitude may be explained by the weak J -couplings between the ^1H and ^{15}N spins, making polarization transfer to be inefficient. Another possible explanation is that the relaxation times of ^1H and ^{15}N spins are substantially different at 500 nT, or during sample transport, but they were not measured here.

The integral intensities of the hyperpolarized ^1H signals correspond to the stoichiometric ratio of protons in $^{15}\text{N}_2$ -ABZ, when SABRE SHEATH is performed at 500 nT. Also, in this case, the detected spin order of protons corresponds to the integral magnetization which changes when SABRE is performed at another magnetic field.

The field dependence of the enhancements produced by SABRE SHEATH is shown in Figure 6. Let us first consider the ^{15}N signals (Figure 6a). The maximal enhancement for the free *cis*- $^{15}\text{N}_2$ -ABZ was achieved in the field range 200–500 nT (Fig-

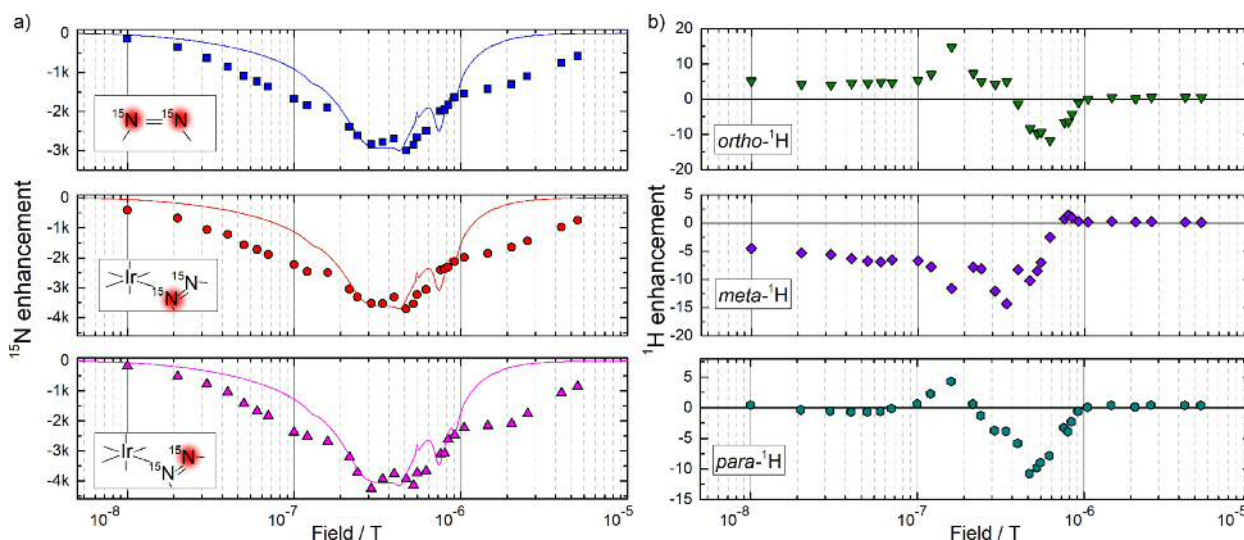


Figure 6. Field dependence of ^{15}N NMR signals (a) and ^1H NMR signals (b) hyperpolarized by SABRE SHEATH. The experimental values of the signal integrals are shown by scatter plots; the calculation results are shown by the solid lines. Details of the calculations are given in the text. All the functions are normalized by the integral intensity of the corresponding signal intensity at thermal equilibrium. Note that the Y-scale is different for each subplot. Top row in (a) shows the data for free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. Middle row in (a) shows the data for the ^{15}N spin of the $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ complex in equatorial position that is directly attached to Ir-IMes. Bottom row in (a) shows the data for the ^{15}N position of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ remote from the Ir-IMes complex. Top row in (b) shows data for the $\text{ortho-}^1\text{H}$ spins in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. Middle row in (b) shows data for the $\text{meta-}^1\text{H}$ spins in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. Bottom row in (b) shows data for the $\text{para-}^1\text{H}$ spins in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. The experimental protocol is shown in Figure 2a.

ure 6a, top row). The field dependences of ^{15}N signals corresponding to $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ in equatorial position of the Ir-IMes complex behave similarly with the signal corresponding to the free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ (compare rows in Figure 6a). This appearance may be explained by the fast chemical exchange of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ between its free form and the form bound in equatorial position: by the time the sample arrives inside the NMR spectrometer (stage iii in Figure 2a) all the $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ molecules might have experienced the chemical exchange many times and the observed polarization of ^{15}N spins in the complex is secondary. It is formed by the hyperpolarized molecules that came from the bulk.

The enhancement factors ε for the ^{15}N signals in the complex are difficult to measure because it would take too much time to acquire thermally polarized signals for them. Therefore, the presented enhancements were estimated using the following procedure: first, the amount of complexes with $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ was determined from the ^1H spectrum. To do that we integrated the signals of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ and the signals corresponding to hydrides at ~ -20 ppm in the ^1H NMR spectrum of thermally polarized sample. This was done for a sample pressurized with 3 bar (same as in SABRE experiments) of thermally polarized H_2 gas. We assume that the ^1H NMR signal of the hydrides at around -20 ppm can be used to quantify the amount of the Ir-IMes complex with $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ bound in the equatorial position (see Supporting Information). The ratio between free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ with respect to $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ bound in the equatorial position determined in such a way was found to be about 34. This ratio was then multiplied by the ratio of the integrals in the ^{15}N spectra of the hyperpolarized signal with respect to the signal of the free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ in

thermal equilibrium and additionally multiplied by a factor of 2 due to stoichiometry. This gave us the enhancements shown in Figure 6a.

This field dependence of ^{15}N signals is reminiscent of the analogous field dependence for other substrates, for example, for ^{15}N -enriched pyridine.^[21] Good agreement is seen between experiments and calculated curves shown as the solid lines in Figure 6a. The experimental function is somewhat broader, probably, due to the presence of additional spins in the system that were not considered in the simulation or due to limited lifetime of the active complex where polarization transfer happens. The calculation was done for a simplified spin system with just 6 spins: 2 ^1H spins coming from the parahydrogen and 4 ^{15}N spins, corresponding to 2 $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ molecules in the equatorial position of the Ir-IMes complex. The calculation was done using the program iRelax available online;^[26] the detailed description of the calculation can be found in reference.^[21] We note that this calculation does not consider the spin dynamics of the polarization transfer, nor the lifetime of the complex. First, the projection of the initial singlet state of parahydrogen onto the eigen states of the full spin system was calculated. All coherences in the obtained density matrix were set to zero and then the Liouville bracket of the obtained state with the \hat{I}_z operator of all ^{15}N spins was taken. Interestingly, when larger spin systems with several ^1H spins of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ were calculated, they showed poor agreement between experimental and calculated data. We believe that this discrepancy is due to the weak polarization levels observed for the ^1H spins. Hence, the hyperpolarization of the ^{15}N magnetization in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ was not strongly affected by the presence of ^1H spins.

The magnetic fields that were applied inside the magnetic shield (Figure 2a, stage ii) were always collinear to the field inside the NMR spectrometer. The observed sign of the ^{15}N signal of the free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ in SABRE SHEATH experiments was always opposite to the thermally polarized one. This observation confirms that the sign of the $^1J_{\text{HH}}$ -coupling between the parahydrogen protons in the Ir-IMes complex is negative: simulation with a positive $^1J_{\text{HH}}$ -coupling inverts the sign of ^{15}N polarization. This is also in agreement with the values for similar complexes, but with different substrates.^[27]

The field dependence of the ^1H spin hyperpolarization in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ is shown in Figure 6b. Typically, ^1H spins of substrates are hyperpolarized by SABRE in a few mT magnetic field.^[27] However, the ^1H spins of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ were hyperpolarized by SABRE only in the ultralow fields. This means that they cannot be directly polarized by the parahydrogen protons in the Ir-IMes complex, but instead their hyperpolarization arises as result of polarization transfer from the hyperpolarized ^{15}N spins. Calculation of the field dependencies of the ^1H spins should take into account propagation of the polarization averaged over the lifetime of the complex; such simulations go beyond the scope of this work. There are two fields at which all curves of the ^1H spins peak: at 150 nT and ~500 nT. Data for the *ortho*-protons look similar to the data of *para*-protons. Both groups change sign of their polarization in the field range from 200 to 400 nT. When the field is lowered to almost zero no net magnetization can be formed. However, some hyperpolarization is still observed; it corresponds to antiparallel orientation between *ortho*- and *meta*-protons. Upon increasing the magnetic field beyond 1 uT no hyperpolarization is generated.

2.2. ^{15}N Singlet Order Formed by SABRE SHEATH

Using the experimental scheme shown in Figure 2b, we acquired NMR spectra detecting only the singlet order of the ^{15}N spin pair of free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. Before the SLIC excitation,^[22] a

so-called T_{00} filter^[23] was applied to destroy all spin orders except for the singlet order. The following SLIC pulse converted hyperpolarized singlet order of the ^{15}N spin pair into either ^{15}N coherence (Figure 7a) or into coherence of the *ortho*-protons of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ (Figure 7b). Due to the T_{00} filter and the selective excitation, these NMR spectra contain only the hyperpolarized signal. A specific spectral feature was observed in the ^{15}N spectrum: the main line was surrounded by weak satellites with the line splitting between them equal to approximately two $^2J_{\text{NN}}$ -couplings. These satellites correspond to enhanced outer singlet-triplet coherences that sometimes can be observed in singlet NMR experiments with nearly equivalent spin pairs.^[28,29]

These experiments clearly show that apart from the ^{15}N and ^1H magnetization, singlet order of the ^{15}N spin pair is populated during SABRE SHEATH experiments. We note that equilibrium population of the singlet order is usually zero at any magnetic field. The observed lines are 400 times (for ^{15}N) or 30 times (for ^1H) stronger than the normal NMR signals.

The field dependence of the hyperpolarization of ^{15}N singlet order was measured via the ^{15}N channel (Figure 8a) and via the ^1H channel (Figure 8b). Both curves shown in Figure 8 have a similar appearance, which is expected as they correspond to two different ways of observing the same hyperpolarized state. Singlet order is formed most efficiently at zero magnetic field; with increasing field the polarization level gradually drops to zero. There is a wide range of magnetic fields (10^{-5} to 10^{-2} T) where the effect is roughly constant. The dependence is somewhat similar to the SABRE experiments on singlet order of the ^{15}N spin pair in diazirines.^[16,17] There is a substantial formation of singlet order observable up to 10^{-2} T. Theis *et al.* explained this feature by observing that singlet order-enrichment happens due to mixing of spin levels that do not depend on the magnetic field.^[16]

We also measured the field dependence of the relaxation times of the hyperpolarized ^{15}N singlet order. The lifetime was below 10 s at every field and thus not longer than the corresponding longitudinal relaxation time of ^{15}N spins in

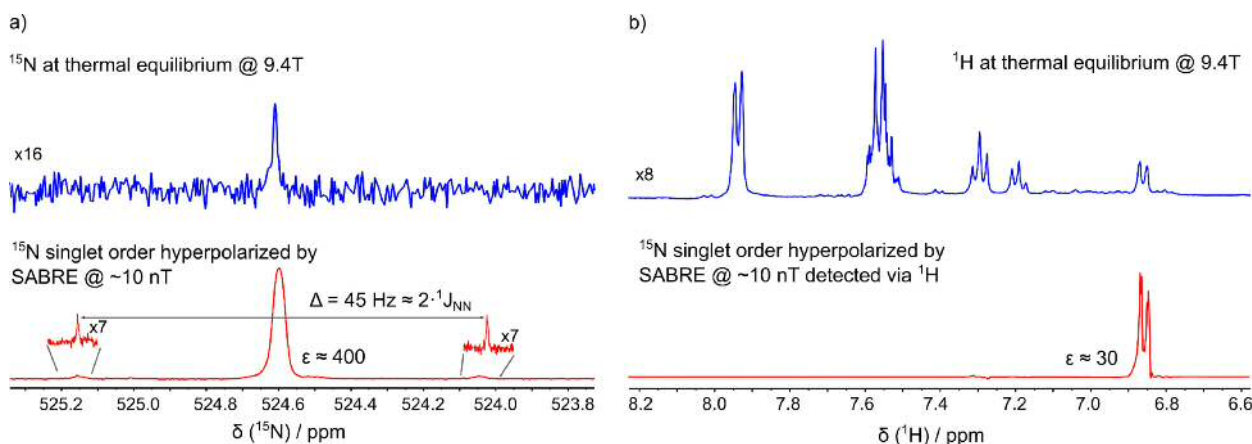


Figure 7. ^{15}N NMR spectra (a) and ^1H NMR spectra (b) showing a comparison of $^{15}\text{N}_2\text{-ABZ}$ signals in thermal equilibrium (blue) and after SABRE SHEATH hyperpolarization at 10 nT of the singlet order of the ^{15}N spin pair (red). The thermal spectrum in (a) was acquired in 4 scans with proton decoupling. This narrowed the line and increased SNR. Hyperpolarized ^{15}N spectrum exhibits outer singlet-triplet coherences. The experimental protocol for the SABRE experiments is shown in Figure 2b.

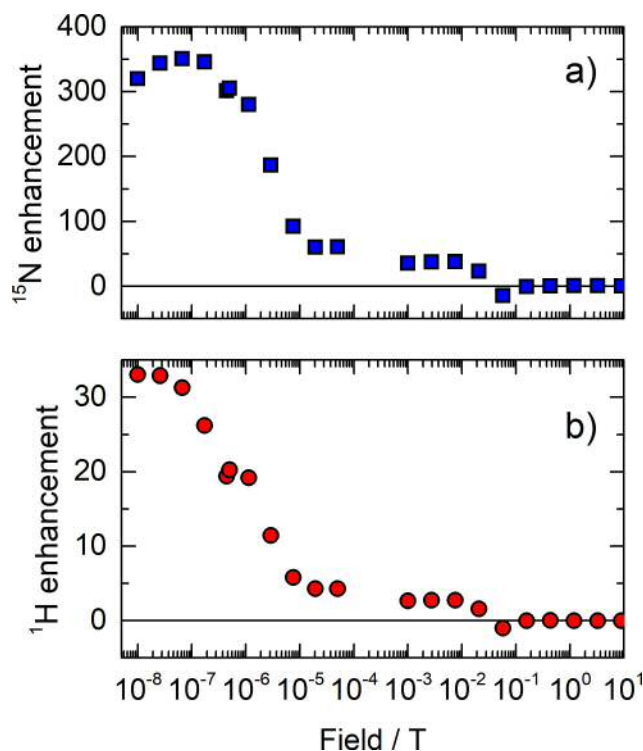


Figure 8. Field dependence of ^{15}N NMR signals (a) and ^1H NMR signals (b) corresponding to singlet order of the ^{15}N spin pair in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ hyperpolarized by SABRE SHEATH. The experimental protocol for the SABRE experiments is shown in Figure 2b. All the functions were normalized by the integral intensity of the corresponding signal intensity at thermal equilibrium.

$\text{cis-}^{15}\text{N}_2\text{-ABZ}$. The absence of long-living character is however expected for $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ and is in accordance with previous observations.^[19] One reason is that $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ experience chemical exchange between the free and bound forms, with the later form having in general much shorter relaxation times and active mechanisms that may cause relaxation of the ^{15}N singlet order. The second reason is that even in the free form the molecular geometry of cis-ABZ is not symmetric, making it possible for dipolar interactions with the protons in the phenyl rings to drive relaxation of the ^{15}N singlet order. This was thoroughly analysed by Stevanato *et al.* for a *cis*- and *trans*-derivative of 2,3- $^{13}\text{C}_2$ -butenedioic acid.^[30]

The enhancement factor for the observation of singlet order via the ^{15}N channel reaches $\epsilon \sim 400$ and ~ 30 via the ^1H channel. A factor of 10 between these values is expected, because ^1H spins in thermal equilibrium have an approximately 10 times higher polarization than ^{15}N spins.

3. Conclusions

In this work we studied SABRE SHEATH experiments of $^{15}\text{N}_2\text{-ABZ}$ and have shown that this molecule is a suitable substrate. There is no need to add other substrates (e.g. pyridine) to the solution to activate the $[\text{Ir}(\text{IMes})(\text{COD})\text{Cl}]$ precatalyst. Addition of *cis*-ABZ

and bubbling of hydrogen is enough. Only $\text{cis-}^{15}\text{N}_2\text{-ABZ}$, but not $\text{trans-}^{15}\text{N}_2\text{-ABZ}$, is hyperpolarized in its free form. On the other hand, we observe complicated ^{15}N spectra hyperpolarized by SABRE SHEATH. There are many hyperpolarized signals corresponding to $^{15}\text{N}_2\text{-ABZ}$ bound to Ir-IMes, and further research is necessary to make the proper assignment of these complexes and to measure their properties. Here, we focussed on studying in detail SABRE SHEATH of free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ and of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ bound to the equatorial position of the Ir-IMes complex.

Our study provides details about optimal fields to perform SABRE SHEATH experiments with $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ in methanol. Further interest in this molecule is caused by the presence of a long-lived state in $\text{trans-}^{15}\text{N}_2\text{-ABZ}$. The lifetime of the singlet order of the ^{15}N spin pair in $\text{trans-}^{15}\text{N}_2\text{-ABZ}$ reaches tens of minutes even at high magnetic fields ($> 10\text{ T}$),^[12] thus making it interesting for high-field MRI applications. Experiments combining SABRE hyperpolarization of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$, its photoswitching to $\text{trans-}^{15}\text{N}_2\text{-ABZ}$ and storage of the hyperpolarization in the LLS of the ^{15}N spin pair are currently underway in our laboratory.

Here we compared SABRE SHEATH experiments where the ^{15}N magnetization is observed (Figure 2a) to those where the ^{15}N singlet order is preserved and subsequently converted into detectable signals (Figure 2b) in $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. The spectral enhancement is in the latter case reduced by a factor of about 10. The optimal field range to polarize ^{15}N and ^1H magnetization lies between 200 and 500 nT, while the optimal magnetic fields to polarize singlet order of the ^{15}N spin pair are in the range between -100 to 100 nT , though even at relatively high magnetic fields of 10^{-2} T some hyperpolarized singlet order can still be generated.

The achievable hyperpolarization level at optimal field that is observed via the ^1H channel is highest for the experiments where singlet order of the ^{15}N spin pair is observed. The enhancement $\epsilon \sim 30$ means that for achieving the same SNR in the thermally polarized ^1H NMR spectrum of $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ it would take about 900 times longer acquisition time. Detection of the hyperpolarized singlet order of the ^{15}N spin pairs via the ^1H channel seems to be an interesting perspective for several reasons. First, the detection via the ^1H channel is more sensitive by ~ 2 orders of magnitude than detection via the ^{15}N channel. Second, this channel is preferable for potential MRI applications, as most of MRI instruments are designed to detect the ^1H magnetization. Finally, we note that the scheme shown in Figure 2b achieves highly selective excitation of the hyperpolarized signals of free $\text{cis-}^{15}\text{N}_2\text{-ABZ}$. This selectivity is achieved through the pulse sequence where first the gradient filter is applied and then selective excitation by SLIC pulse is performed. This allows observation of only the hyperpolarized component, thus making $\text{cis-}^{15}\text{N}_2\text{-ABZ}$ an interesting molecule, that can be used in singlet-contrast MRI experiments.^[31]

Supporting Information

In addition to the pdf file provided with this manuscript, also some NMR spectra are available to download: <https://github.com/kirillshes/SABRE-Azobenzene-SI-data>

Acknowledgments

K.F.S. is grateful to James Eills and Danila Barskiy for useful discussions and help with editing of the manuscript. We acknowledge the Russian Science Foundation for the grant #20-63-46034. K.F.S. has received funding from the European Union's Horizon 2020 research and innovation programme under the Marie Skłodowska-Curie grant agreement No 766402. Open access funding enabled and organized by Projekt DEAL.

Conflict of Interest

The authors declare no conflict of interest.

Keywords: Parahydrogen · NMR · SABRE · azobenzene · ZULF

- [1] P. J. Rayner, M. J. Burns, A. M. Olaru, P. Norcott, M. Fekete, G. G. R. Green, L. A. R. Highton, R. E. Mewis, S. B. Duckett, *PNAS* **2017**, DOI 10.1073/pnas.1620457114.
- [2] M. Fekete, F. Ahwal, S. B. Duckett, *J. Phys. Chem. B* **2020**, *124*, 4573–4580.
- [3] R. W. Adams, J. A. Aguilar, K. D. Atkinson, M. J. Cowley, P. I. P. Elliott, S. B. Duckett, G. G. R. Green, I. G. Khazal, J. López-Serrano, D. C. Williamson, *Science* **2009**, *323*, 1708–1711.
- [4] J. F. P. Colell, A. W. J. Logan, Z. Zhou, R. V. Shchepin, D. A. Barskiy, G. X. Ortiz, Q. Wang, S. J. Malcolmson, E. Y. Chekmenev, W. S. Warren, T. Theis, *J. Phys. Chem. C* **2017**, *121*, 6626–6634.
- [5] J.-B. Hövener, A. N. Pravdivtsev, B. Kidd, C. R. Bowers, S. Glöggler, K. V. Kovtunov, M. Plaumann, R. Katz-Brull, K. Buckenmaier, A. Jerschow, F. Reineri, T. Theis, R. V. Shchepin, S. Wagner, P. Bhattacharya, N. M. Zacharias, E. Y. Chekmenev, *Angew. Chem. Int. Ed. Engl.* **2018**, *57*, 11140–11162.
- [6] D. A. Barskiy, S. Knecht, A. V. Yurkovskaya, K. L. Ivanov, *Prog. Nucl. Magn. Reson. Spectrosc.* **2019**, *114–115*, 33–70.
- [7] T. Theis, M. L. Truong, A. M. Coffey, R. V. Shchepin, K. W. Waddell, F. Shi, B. M. Goodson, W. S. Warren, E. Y. Chekmenev, *J. Am. Chem. Soc.* **2015**, *137*, 1404–1407.
- [8] M. L. Truong, T. Theis, A. M. Coffey, R. V. Shchepin, K. W. Waddell, F. Shi, B. M. Goodson, W. S. Warren, E. Y. Chekmenev, *J. Phys. Chem. C* **2015**, *119*, 8786–8797.
- [9] M. H. Levitt, *J. Magn. Reson.* **2019**, *306*, 69–74.
- [10] G. Pileio, M. Carravetta, M. H. Levitt, *PNAS* **2010**, *107*, 17135–17139.
- [11] S. J. Elliott, P. Kadeřávek, L. J. Brown, M. Sabba, S. Glöggler, D. J. O'Leary, R. C. D. Brown, F. Ferrage, M. H. Levitt, *Mol. Phys.* **2019**, *117*, 861–867.
- [12] K. F. Sheberstov, H.-M. Vieth, H. Zimmermann, B. A. Rodin, K. L. Ivanov, A. S. Kiryutin, A. V. Yurkovskaya, *Sci. Rep.* **2019**, *9*, 20161.
- [13] S. S. Roy, P. Norcott, P. J. Rayner, G. G. R. Green, S. B. Duckett, *Angew. Chem. Int. Ed.* **2016**, *55*, 15642–15645; *Angew. Chem.* **2016**, *128*, 15871–15874.
- [14] S. S. Roy, P. J. Rayner, M. J. Burns, S. B. Duckett, *J. Chem. Phys.* **2020**, *152*, 014201.
- [15] G. Stevanato, J. T. Hill-Cousins, P. Håkansson, S. S. Roy, L. J. Brown, R. C. D. Brown, G. Pileio, M. H. Levitt, *Angew. Chem. Int. Ed.* **2015**, *54*, 3740–3743; *Angew. Chem.* **2015**, *127*, 3811–3814.
- [16] T. Theis, G. X. Ortiz, A. W. J. Logan, K. E. Claytor, Y. Feng, W. P. Huhn, V. Blum, S. J. Malcolmson, E. Y. Chekmenev, Q. Wang, W. S. Warren, *Sci. Adv.* **2016**, *2*, e1501438.
- [17] K. Shen, A. W. J. Logan, J. F. P. Colell, J. Bae, G. X. Ortiz, T. Theis, W. S. Warren, S. J. Malcolmson, Q. Wang, *Angew. Chem. Int. Ed. Engl.* **2017**, *56*, 12112–12116.
- [18] I. Kownacki, M. Kubicki, K. Szubert, B. Marciniec, *J. Organomet. Chem.* **2008**, *693*, 321–328.
- [19] K. F. Sheberstov, H.-M. Vieth, H. Zimmermann, K. L. Ivanov, A. S. Kiryutin, A. V. Yurkovskaya, *Appl. Magn. Reson.* **2018**, *49*, 293–307.
- [20] J. L. Markley, A. Bax, Y. Arata, C. W. Hilbers, R. Kaptein, B. D. Sykes, P. E. Wright, K. Wüthrich, *Eur. J. Biochem.* **1998**, *256*, 1–15.
- [21] A. S. Kiryutin, A. V. Yurkovskaya, H. Zimmermann, H.-M. Vieth, K. L. Ivanov, *Magn. Reson. Chem.* **2018**, *56*, 651–662.
- [22] S. J. DeVience, R. L. Walsworth, M. S. Rosen, *Phys. Rev. Lett.* **2013**, *111*, 173002.
- [23] M. C. D. Tayler, in *Long-Lived Nuclear Spin Order*, The Royal Society Of Chemistry, **2020**, pp. 188–208.
- [24] R. V. Shchepin, L. Jaigirdar, T. Theis, W. S. Warren, B. M. Goodson, E. Y. Chekmenev, *J. Phys. Chem. C* **2017**, *121*, 28425–28434.
- [25] L. Carlton, M.-P. Belciug, *J. Organomet. Chem.* **1989**, *378*, 469–472.
- [26] Relax program can be found at: <http://www.tomo.nsc.ru/en/nmr/iRelax/>.
- [27] A. N. Pravdivtsev, K. L. Ivanov, A. V. Yurkovskaya, P. A. Petrov, H.-H. Limbach, R. Kaptein, H.-M. Vieth, *J. Magn. Reson.* **2015**, *261*, 73–82.
- [28] K. F. Sheberstov, A. S. Kiryutin, C. Bengs, J. T. Hill-Cousins, L. J. Brown, R. C. D. Brown, G. Pileio, M. H. Levitt, A. V. Yurkovskaya, K. L. Ivanov, *Phys. Chem. Chem. Phys.* **2019**, *21*, 6087–6100.
- [29] J. Eills, J. W. Blanchard, T. Wu, C. Bengs, J. Hollenbach, D. Budker, M. H. Levitt, *J. Chem. Phys.* **2019**, *150*, 174202.
- [30] G. Stevanato, S. S. Roy, J. Hill-Cousins, I. Kuprov, L. J. Brown, R. C. D. Brown, G. Pileio, M. H. Levitt, *Phys. Chem. Chem. Phys.* **2015**, *17*, 5913–5922.
- [31] J. Eills, E. Cavallari, R. Kircher, G. D. Matteo, C. Carrera, L. Dagys, M. H. Levitt, K. L. Ivanov, S. Aime, F. Reineri, K. Münnemann, D. Budker, G. Buntkowsky, S. Knecht, *Angew. Chem. Int. Ed.* **2021**, *60*, 6791–6798; *Angew. Chem.* **2021**, *133*, 6866–6873.

Manuscript received: February 27, 2021
Revised manuscript received: April 2, 2021
Accepted manuscript online: May 1, 2021
Version of record online: June 8, 2021